

**unit 6 chemistry test pdf**

6. If 3.0 liters of oxygen gas are combined with 4.0 liters of hydrogen gas, how many moles of water will form at STP?  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$  a. 0.13 mol b. 0.18 mol c. 0.27 mol d. 0.36 mol 7. How many grams of magnesium phosphate are needed to make 250. mL of a 0.235 M solution? a. 6.99 g b. 9.81 g c. 15.4 g d. 19.7 g 8.

**Unit 6 Test - William and Mary**

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**Unit 6 - Quantitative Chemistry - Mr. Grayson's Honors**

Page 3 of 6. 21. Which of the following household chemicals is an acid? (A) bleach (B) drain cleaner (C) vinegar (D) all of the above 22. Which of the following could be used to determine whether or not a solution is an acid? (A) red litmus paper (B) blue litmus paper (C) both (A) and (B) (D) neither (A) nor (B) 23.

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A)  $6.05 \times 10^{-3}$  B)  $2.74 \times 10^{-19}$  C)  $4.56 \times 10^{-46}$  D) 227 E) 165 6) Of the following transitions in the Bohr hydrogen atom, the \_\_\_\_\_ transition results in the emission of the highest-energy photon.

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Honors Chemistry Unit 6 Practice Test (Periodic Table and Trends) Matching Match each item with the correct statement below. a. group d. metal b. ionization energy e. periodic law c. atomic radius \_\_\_\_ 1. vertical column in the periodic table \_\_\_\_ 2. A repetition of properties occurs when elements are arranged in order of increasing atomic number. \_\_\_\_ 3.

## **Unit 6 Practice Test (Periodic Table) (1) - Honors**

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